THALLIUM NITRATE AS A REAGENT FOR THE CONVERSION OF EPOXIDES INTO d-HYDROXY-

NITRATE ESTERS AND FOR THE CLEAVAGE OF ALTPHATIC ETHERS E. Mincione and F. Lanciano

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The efficient and versatile reactivity of thallium(III) nitrate with ketones and alkenes is well known (1). We reported that this salt is also able to react with weakly nucleophilic functional groups such as epoxides and aliphatic ethers. In fact, simple as well steroidal oxiranes are converted at room temperature, in acetic acid or hexane solution, on treatment with a molar equivalent of $Tl(NO_3)_3$, into the corresponding α -hydroxy nitrate esters (Table 1). In a similar way, aliphatic ethers, in acetic anhydride solution, undergoe cleavage of the ethereal bond with the formation of the corresponding acetoxy derivatives.

TABLE 1. Reaction of epoxides and aliphatic ethers with $Tl(NO_3)_3$

Starting materials	solvent	Products, yield %		
	acet1c	cyclohexanediol mono-	cyclohexanediol mono-	
cyclohexene oxıde I	acıd	nıtrate ester II (62)	acetate III (30)	
	hexane	II (76)		
styrene oxıde IV	acetic	1-pheny1-2-hydroxy-	1-phenyl-2-hydroxy-1-	
	acıd	1-ethyl nitrate V (40)	ethyl acetate VI (44)	
	hexane	V (65)		
2 α,3α- chole- stane oxide VII	acet1c	3 a -hydroxy-5 a -cholestan-	3 α- hydroxy-5 α- cholestan-	
	acıd	2 β -yl nitrate VIII (57)	2 $meta$ -yl acetate IX (21)	
	methanol	3 α -hydroxy -5α -cholestan- 2β-yl methyl ether X (76)		
	hexane	VIII (80)		
3β -acetoxy-5 $\alpha, 6\alpha$ -cholesta- ne oxide XI	acetic acid	3β -acetoxy- 5α -hydroxy- cholestan- 6β -yl nitrate XII (75)	5α -hydroxy- 3β , 6β -diaceto- xy cholestane XIII (21)	
	hexane	XII (81)		
3β -acetoxy-5 β ,6 β -cholesta- ne oxide XIV	acetic acid	3 β- acetoxy-6 β- hydroxy- cholestan-5 α- yl nitrate XV (25)	6β -hydroxy- 3β ,5 α -diaceto- xy cholestane XVI (57)	
	hexane	XV (68)		

3 β -methoxy- cholestane	acetıc anhydrıde	3 5 -acetoxy-cholestane (91)
2-methoxy-trans- decalın	"	2-acetoxy-trans-decalın (68)
dibutyl ether	**	butyl acetate (57)
tetrahydrofuran	**	4-acetoxybutan-1-ol (62)

As regards the stereochemical course of the oxirane ring opening in the steroidal compounds (VII) (XI) (XIV), the formation of the trans diol mononitrate, as expected, is observed.

The nitrate esters are partly transesterified in the reaction medium to the corresponding acetoxy derivatives, with retention of configuration. This acetolysis is complete if the reaction temperature is raised to 80-100°C. Moreover, by examination of solvent effects on the reaction course we conclude that in hexane solution the reaction proceeds to give the best yields of nitrate esters (Table 1), although with a slower conversion rate, while in methanol solution the formation of monomethoxy diols such as (X) is observed. As regards the influence of the salt anion on the effectiveness of the reaction, by utilising a thallium salt with a less electronegative anion, such as the thallium (III) acetate, the opening reaction proceeds very slowly and in low yield (20%). Hydroxy nitrate esters appear to be very unstable and reactive compounds; in order to isolate them, by the initially reported method (2), very mild and convenient conditions are necessary.

As to the effectiveness on the ethereal linkage, while no reaction is observed by thallium nitrate in acetic acid, also after prolonged warming of the ethereal solution at 50-80°C, a effective breaking of the aliphatic ethereal bond takes place even at room temperature in acetic anhydride solution by a molar excess of $T1(NO_3)_3$, with subsequent formation of the corresponding acetoxy derivatives (Table 1.) This is a useful and promising result if we consider the severe conditions required for the cleavage of the ethereal linkage.

NOTES AND BIBLIOGRAPHY

- 1. A. McKillop and E.C. Taylor, Endeavour, 88-93 (1976).
- 2. Formation of α -nitratoketones from the reaction of enolisables ketones with $TI(NO_3)_3$ (3) and formation of nitrate esters from reaction of alkenes with $TI(NO_2)_2$ (4) has been reported recently.
- 3. A. McKillop, D.W. Young, M. Edwards, R.P. Hug and E.C. Taylor, <u>J.Org.Chem</u>., <u>43</u>, 3773 (1978).
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